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## Key indicators

Single-crystal X-ray study

T = 294 K

Mean  $\sigma(\text{C}-\text{C}) = 0.003 \text{ \AA}$ 

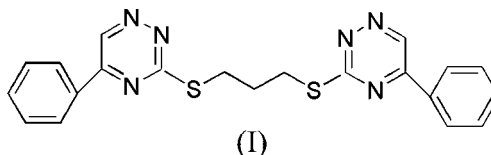
R factor = 0.044

wR factor = 0.126

Data-to-parameter ratio = 15.3

For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.5-Phenyl-3-[3-(5-phenyl-1,2,4-triazin-3-ylsulfanyl)-  
propylsulfanyl]-1,2,4-triazineThe title compound,  $\text{C}_{21}\text{H}_{18}\text{N}_6\text{S}_2$ , crystallizes with one half-  
molecule in the asymmetric unit; a twofold rotation axis passes  
through the central C atom. The triazine ring makes a dihedral  
angle of  $10.65(2)^\circ$  with the benzene ring.Received 1 September 2006  
Accepted 21 September 2006

## Comment

1,2,4-Triazine derivatives are an important class of compounds  
that exhibit various biological activities (Konno *et al.*, 1995)  
and pharmacological properties (Monge *et al.*, 1991). We  
synthesized the title compound, (I), as part of our systematic  
studies aimed at developing new bioactive compounds.The structure has one half-molecule in the asymmetric unit  
with the molecule sitting on a twofold axis passing through  
atom C11 (Fig. 1). All atoms in the triazine ring are essentially  
coplanar; the largest deviation from the mean plane is  
 $0.031(2) \text{ \AA}$  for atom C9. The dihedral angle between the  
triazine ring and the benzene ring is  $10.65(2)^\circ$ . Bond distances  
and angles (Table 1) of the triazine ring are as expected for  
this type of compound (Wen *et al.*, 2006). The molecules pack  
in staggered layers stabilized mainly by van der Waals forces.

## Experimental

1,4-Dibromopropane (0.404 g, 2 mmol) was added to a solution of 5-  
phenyl-1,2,4-triazine-3-thiol (0.756 g, 4 mmol) in aqueous ethanol  
(10 ml, 50%) containing KOH (0.224 g, 4 mmol) and tetraethyl-  
ammonium bromide (0.05 g). The reaction mixture was refluxed for  
6 h and then poured into water (100 ml). The resulting precipitate  
was collected and crystallized from mixed solvents, hexanes and ethyl  
acetate (2:1), to give the title compound. Single crystals suitable for  
X-ray analysis were obtained as colorless blocks by slow evaporation  
from ethyl acetate (m.p.  $>503 \text{ K}$ , decomposition).

## Crystal data

 $\text{C}_{21}\text{H}_{18}\text{N}_6\text{S}_2$  $M_r = 418.53$ Monoclinic,  $C2/c$  $a = 18.204(8) \text{ \AA}$  $b = 11.495(5) \text{ \AA}$  $c = 9.692(5) \text{ \AA}$  $\beta = 102.483(8)^\circ$  $V = 1980.2(16) \text{ \AA}^3$ 

Z = 4

 $D_x = 1.404 \text{ Mg m}^{-3}$ Mo  $K\alpha$  radiation $\mu = 0.29 \text{ mm}^{-1}$ 

T = 294(2) K

Block, colorless

 $0.22 \times 0.12 \times 0.10 \text{ mm}$

## Data collection

Bruker APEX-II CCD area-detector diffractometer  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)  
 $T_{\min} = 0.939$ ,  $T_{\max} = 0.972$

5456 measured reflections  
 2014 independent reflections  
 1231 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.044$   
 $\theta_{\text{max}} = 26.3^\circ$

## Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.044$   
 $wR(F^2) = 0.126$   
 $S = 0.95$   
 2014 reflections  
 132 parameters

H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.059P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.30 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.18 \text{ e } \text{\AA}^{-3}$

Table 1

Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ ).

S1—C9	1.745 (2)	N3—C7	1.321 (3)
S1—C10	1.805 (2)	N3—C9	1.347 (3)
N1—C9	1.319 (3)	C7—C8	1.399 (3)
N1—N2	1.350 (3)	C11—C10 <sup>i</sup>	1.517 (3)
N2—C8	1.313 (3)		
C9—S1—C10	103.16 (12)	N2—C8—C7	122.7 (2)
C9—N1—N2	116.81 (18)	N1—C9—N3	127.3 (2)
C7—N3—C9	115.3 (2)	C10 <sup>i</sup> —C11—C10	111.0 (3)
N3—C7—C8	118.9 (2)		

Symmetry code: (i)  $-x + 1, y, -z + \frac{3}{2}$ .

All H atoms were placed in calculated positions, with C—H = 0.93 or 0.97  $\text{\AA}$ , and included in the final cycles of refinement using a riding model, with  $U_{\text{iso}}(\text{H})$  set at  $1.2U_{\text{eq}}(\text{C})$  for all H atoms.

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve

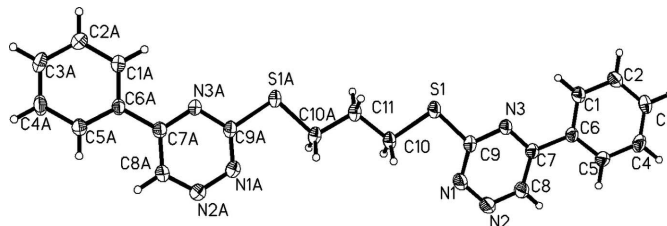


Figure 1

The molecular structure, with displacement parameters at the 35% probability level. The suffix A corresponds to symmetry code i in Table 1.

structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1999); software used to prepare material for publication: SHELXTL.

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